Hydrochloric Acid (HCL) 30-32%

Hydrochloric Acid (HCl) is a colorless to yellowish-green, clear corrosive liquid with a pungent, irritating odour.

It is completely soluble in water and Hydrochloric Acid gas mixed with water generate a violent exothermique reaction. Contact with most metals results in the formation of flammable hydrogen gas.

Hydrochloric Acid is used in numerous applications from water treatment to food processing.

Properties

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Molecular Weight: 36.5g/mol
Boiling Point: 80^{\circ}C(@conc.32\%); 45C@conc.37\%)
Freezing Point: -42^{\circ}C(@conc.32\%); -29^{\circ}@conc.37\%)
Vapor Pressure (at 20^{\circ}C): 30hPa, at20^{\circ}(Concentration:32\%), 200hPa, at20^{\circ}(Concentration:37\%)
Liquid Density (Water = 1.0): 1.16 - 1.19kg/m3 (@20^{\circ}C)
Vapor Density (Air = 1.0): 1.53kg/m3
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Handling and Storage

Hydrochloric Acid is corrosive. It should be stored away from oxidizing agents and alkaline materials in a cool, dry, ventilated place. Although stable under normal conditions, Hydrochloric Acid reacts with metals to form explosive and flammable hydrogen gas. Adding water to Hydrochloric Acid produces a violent, exothermic reaction.

When fighting fires or acid spills, proper protective clothing and respiratory equipment should be utilized and unprotected personnel should be moved upwind of the area.





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General Information

Technical

Other commercial names:	Hydronium chloride, Hydrogen chloride, Muriatic acid, Chlorhydric Acid
Chemical formula :	HCI
Physical property:	Hydrochloric acid is a clear, pale, and highly odorous solution of hydrogen chloride (HCl) in water
	A Y A
Product type:	Industrial
Interactions:	It is a highly corrosive mineral acid and releases hydrogen in contact with some metals (flammable)
Production Process:	It can also be produced through the following methods: A. From chlorine and hydrogen / B. From electrolysis of salt
Storage:	Hydrochloric acid is usually shipped in steel tankers with antioxidant coatings or special types of plastic.
l Information	
molecular mass	36.46 g/mol
Density	1.18 g/cm ³
melting point (38% solut	ion) -27.32 °C
Boiling point (38% solut	ion) 48 °C
Acidic pKa	-8.0

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Packing

jerry can 25 Lit., 30 Lit.



Drums 220 Lit

IBC Tank 1000 Lit



